

SYNTHESIS OF HETERO-BICYCLIC COMPOUNDS

Part-IX. Formation of 2,2-disubstituted 4,5-dioxo-pyridino [4,3-d] [1,3]dioxins

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Chloropyranodioxins (I) prepared by the reaction of malonyl chloride with mixed ketone were converted into 7-(phenylamino)-pyranodioxins (II) which underwent the phenoxide rearrangement to yield the corresponding pyridinodioxins (III), whose structures were determined by chemical conversions and spectroscopic studies.

Key words: Chloropyranodioxin, Pyridinodioxin, Hetero-bicyclic compounds.

Introduction

Several methods for the synthesis of substituted pyridines have been reported in literature [1-10]. The authors have demonstrated synthetic utility of chloropyranodioxin (I) in their preceding papers of the series [11,12].

In continuation of the previous work [12], preparation of 2, 2-disubstituted-4, 5-dioxo-7-(phenylamino)-pyranodioxins (II) from chloro compounds (I) and their subsequent rearrangement to the corresponding 2,2-disubstituted-4, 5-dioxo-pyridinodioxins (III) under the influence of sodium phenoxide in phenol have been investigated. The characterization of these compounds is accomplished by spectroscopic studies supported by elemental analysis and formation of derivatives.

Experimental

Melting points were determined with a Thomas-Hoover Capillary apparatus and are uncorrected. UV and IR spectra were recorded on Beckman 36 and Perkin Elmer 283 spectrophotometer respectively.

7-Chloro-2-ethyl-2-methyl-4, 5-dioxopyrano [4,3-d] [1,3]dioxin (I, R' = CH₃, R'' = C₂H₅):- A mixture of malonyl chloride (9.7ml, 0.1 mole) and methyl ethyl ketone (4.5 ml, 0.05 mole) was heated on a water bath until a solid product appeared. Trituration of the product with ether gave 7-chloro-2-ethyl-2-methyl-4,5-dioxopyrano[4,3-d] [1,3] dioxin (8.5g, 70%) (I, R' = CH₃, R'' = C₂H₅) which was crystallized from benzene (m.p. 160°C). Found: C, 49.4; H, 3.5. C₁₀H₉O₅Cl requires C, 49.1; H, 3.7%.

Other chloropyranodioxins prepared similarly are listed in Table 1.

2-Ethyl-2-methyl-4,5-dioxo-7-(phenylamino)-pyrano [4,3-d] [1,3] dioxin (R' = CH₃, R'' = C₂H₅):- To a solution of

7-chloro-2-ethyl-2-methyl-4, 5-dioxo-pyrano [4,3-d] [1,3] dioxin (4.9g, 0.02 moles) (I, R' = CH₃, R'' = C₂H₅) in CHCl₃ (10 ml) aniline (3.7g; 0.04 mole) in CHCl₃ (10ml) was added with constant stirring. A solid product was obtained which was washed with water and dried. 2-Ethyl-2-methyl-4, 5-dioxo-7-(phenylamino)-pyrano [4,3-d] [1,3] dioxin (4.9g, 82%) (II, R' = CH₃, R'' = C₂H₅) was crystallized from a mixture of CHCl₃ and CH₃OH (m.p. 163°C). Found: C, 63.8; H, 4.9; N, 4.5. C₁₆H₁₅NO₅ requires C, 63.8; H, 4.9; N, 4.6%.

Other 2,2-disubstituted -4,5-dioxo-7-(phenylamino)-pyrano [4,3-d] [1,3] dioxins (II) were prepared as above and are listed in Table 2.

2-Ethyl-2-methyl-4, 5-dioxo-7-hydroxy-6-phenylpyridino [4,3-d] [1,3] dioxin (III, R' = CH₃, R'' = C₂H₅):- 2-Ethyl-2-methyl-4,5-dioxo-7-(phenylamino)-pyrano [4,3-d] [1,3]dioxin (II, R' = CH₃, R'' = C₂H₅) (3g, 0.01 moles) was added to a solution of Na (0.92g, 4 moles) in phenol (30 ml) and the mixture was heated at 120°C for 2 minutes. The solution was cooled, diluted with water and extracted with ether to recover excess phenol. The ethereal layer was extracted with water and the combined aqueous extracts (130 ml) were acidified with HCl (3N) to yield a solid product. The product, 2-ethyl-2-methyl-4,5-dioxo-7-hydroxy-6-phenyl-pyridino [4,3-d][1,3] dioxin (III, R' = CH₃, R'' = C₂H₅; 2.6g, 87%) was crystallized from MeOH, mp 210°C. It gave a reddish colouration with aq. FeCl₃ and showed effervescence with aq. NaHCO₃. Found: C, 63.8; H, 4.8; N, 4.5. C₁₆H₁₅NO₅ requires C, 63.8; H, 4.9; N, 4.6%.

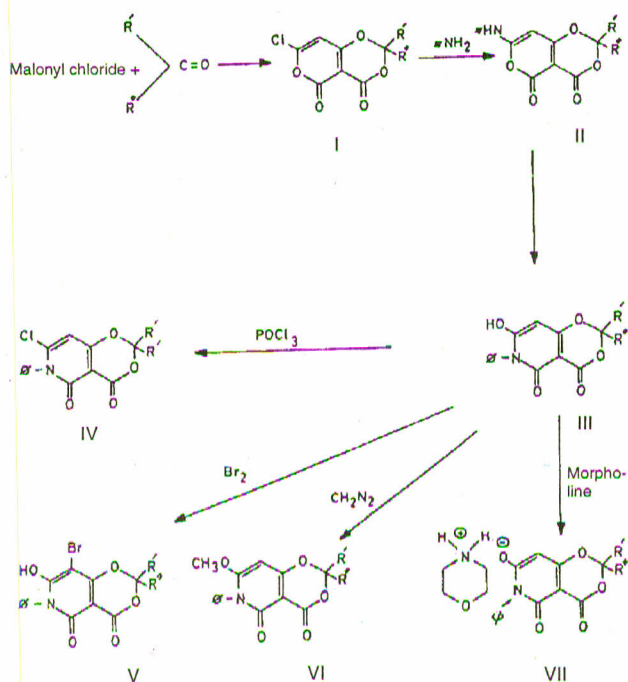
The rearrangement products III of other arylamino-2, 2-disubstituted-pyranodioxins with sodium phenoxide in phenol are listed in Table 3.

7-Chloro-2-ethyl-2-propyl-4, 5-dioxo-6-phenyl-pyridino [4,3-d] [1,3] dioxin (IV, R' = C₂H₅, R'' = C₃H₇):- 2-Ethyl-7-hydroxy-4, 5-dioxo-6-phenyl-2-propyl-pyridino [4,3-d][1,3] dioxin (III, R' = C₂H₅, R'' = C₃H₇; 0.5g) and POCl₃ (10ml) were

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TABLE 1. FORMATION OF 2,2-DISUBSTITUTED-7-CHLORO-4,5-DIOXOPYRANO [4,3-d] [1,3] DIOXINS (I).

No. Keto compound	Malonyl chloride	2,2-disubstituted-7-chloro-dioxo-pyrano [4,3-d][1,3] dioxins (I)	Yield g.	m.p. C°	Molecular Formula	Analysis			
						Found C	H	Required C	H
1. CH ₃ COC ₂ H ₅ (4.5ml)	9.7 ml	7-chloro-2-ethyl-2-methyl-4,5-dioxopyranodioxin	1.6	160	C ₁₀ H ₉ O ₅ Cl	49.0	3.6	49.0	3.7
2. C ₆ H ₅ COCH ₃ (2.5g)	4.0 ml	7-chloro-2-methyl-2-phenyl-4,5-dioxopyranodioxin	3.0	147	C ₁₄ H ₉ O ₅ Cl	57.5	3.0	57.4	3.1
3. C ₆ H ₅ COC ₂ H ₅ (5.6g)	8.0 ml	7-chloro-2-ethyl-2-phenyl-4,5-dioxopyranodioxin	10.0	105	C ₁₅ H ₁₁ O ₅ Cl	58.8	3.6	58.7	3.6
4. p-Cl-C ₆ H ₄ COCH ₃ (6.4g)	8.0 ml	7-chloro-2-methyl-2-(p-chlorophenyl)-4,5-dioxopyranodioxin	7.0	129	C ₁₄ H ₈ O ₅ Cl ₂	51.6	2.2	51.4	2.4
5. C ₆ H ₅ CO.COC ₆ H ₅ (8.7g)	8.0 ml	7-chloro-2-phenyl-2-benzoyl-4,5-dioxopyranodioxin	7.8	88	C ₂₀ H ₁₁ O ₆ Cl	62.8	2.7	62.7	2.8
6. CH ₃ CO(CH ₂) ₃ CH ₃ (2.6ml)	4.0 ml	2-butyl-7-chloro-2-methyl-4,5-dioxopyranodioxin	2.5	91	C ₁₂ H ₁₃ O ₅ Cl	52.9	4.5	52.8	4.7
7. C ₂ H ₅ CO(CH ₂) ₂ CH ₃ (2.6ml)	4.0 ml	7-chloro-2-methyl-2-propyl-4,5-dioxopyranodioxin	3.1	110	C ₁₂ H ₁₃ O ₅ Cl	52.8	4.6	52.8	4.7
8. CH ₃ (CH ₂) ₂ CO(CH ₂) ₂ CH ₃ (3.0ml)	4.0 ml	7-chloro-2,2-dipropyl-4,5-dioxopyranodioxin	3.0	135	C ₁₃ H ₁₅ O ₅ Cl	54.6	5.1	54.4	5.2



heated on a water bath for 30 minutes. Excess of POCl₃ was removed under reduced pressure and the residue was crystallized from EtOH/water (0.2g, 38.5%), m.p. 160°C. Found; C, 62.0; H, 5.1; N, 3.8. C₁₂H₁₃O₅Cl requires C, 62.1; H, 5.2; N, 4.0%.

8-Bromo-2-ethyl-7-hydroxy-4,5-dioxo-2,6-diphenylpyridino-[4,3-d] [1,3]dioxin (V, R'=C₆H₅, R''=C₂H₅): To a solution of the compound (III, R'=C₆H₅, R''=C₂H₅ 0.5g) in CHCl₃ (20 ml), bromine in CHCl₃ was added drop wise till an orange colour persisted. The reaction mixture was kept at

room temperature for 1 h. and the solvent was removed under reduced pressure. A solid bromo product (V, R'=C₆H₅; R''=C₂H₅; 0.35g) was obtained and recrystallized from MeOH. m.p. 124°C. Found: C, 57.7; H, 3.5. C₂₁H₁₆NO₅ Br requires C, 57.0; H, 3.6; N, 3.2%.

2-Butyl-7-methoxy-2-methyl-4, 5-dioxo-6-phenylpyridino [4,3-d] [1,3] dioxin (VI, R'=CH₃, R''=C₄H₉): To 2-butyl-7-hydroxy-2-methyl-4, 5-dioxo-6-phenyl-pyridino [4,3-d] [1,3] dioxin (III, R'=CH₃, R''=C₄H₉; 0.5g) in ether, was added a solution of diazomethane in ether until yellow color persisted. The solution was kept for 2 h. in a refrigerator and the solvent was removed. The residue yielded a product on trituration with ether. The product 2-butyl-7-methoxy-2-methyl-4, 5-dioxo-6-phenyl-pyridino [4,3-d] [1,3] dioxin (VI R'=CH₃, R''=C₄H₉; 0.23g) was crystallized from MeOH, mp 149°C. Found: C,66.5; H,5.9; N,4.0. C₁₉H₂₁NO₅ requires C,66.5; H,6.1; N,4.1%.

Morpholinium salt (VII, R'=C₂H₅ and R''=C₃H₇) of 7-hydroxy product (III, R'=C₂H₅, R''=C₃H₇): To the compound (0.5g) (III, R'=C₂H₅, R''=C₃H₇) in CHCl₃ (10ml) was added morpholine (0.5 ml) and the mixture was refluxed for 3 minutes. The solvent was removed under reduced pressure and the residue obtained (0.5g, 79%) was recrystallized from MeOH, m.p. 193°C. Found C, 63.5; N, 6.7. C₂₂H₂₈N₂O₆ requires C, 63.4; N, 6.7%.

Results and Discussion

We have studied the formation of some hitherto unreported chloropyranodioxins (I) by the reaction of malonyl chloride with mixed ketones. These chloropyranodioxins (I) reacted with aniline to yield 2,2-disubstituted-7-(phe-

TABLE 2. FORMATION OF 2,2-DISUBSTITUTED 4,5-DIOXO-7-(PHENYLAMINO)-PYRANO [4,3-d][1,3] DIOXINS.

No.	2,2-disubstitute-7-chloro-4,5-dioxo-pyranodioxin(I)		Quantity (g)	Aniline (g)	2,2-disubstituted-4,5-dioxo-7-(phenylamino) pyrano-[4,3-d][1,3] dioxin(II)	Yield %	m.p. C°	Molecular Formula	Analysis						UV Light Absorption 95% MeOH	
	R'	R''							Found			Required			λ_{max} (nm)	Loge
									C	H	N	C	H	N		
1.	CH ₃ -	C ₂ H ₅ -	4.9	3.7	2-ethyl-2-methyl-4,5-dioxo-7-(phenylamino) pyranodioxin	82	163	C ₁₆ H ₁₅ NO ₅	63.8	4.9	4.6	63.8	4.8	4.6	341	4.72
2.	CH ₃ -	C ₆ H ₅ -	7.0	4.4	2-methyl-2-phenyl-4,5-dioxo-7-(phenylamino) pyranodioxin	84	147	C ₂₀ H ₁₅ NO ₅	68.8	4.0	3.9	68.7	4.3	4.0	344	4.48
3.	C ₂ H ₅ -	C ₆ H ₅ -	8.0	4.7	2-ethyl-2-phenyl-4,5-dioxo-7-(phenylamino)-pyranodioxin	78	156	C ₂₁ H ₁₇ NO ₅	69.9	4.4	3.8	69.9	4.6	3.8	344	4.52
4.	CH ₃ -	p-Cl-C ₆ H ₄ -	6.0	3.4	2-(p-chlorophenyl)-2-methyl-4,5-dioxo-7-(phenylamino)-pyranodioxin	91	160	C ₂₀ H ₁₄ NO ₅ Cl	62.4	3.3	3.4	62.5	3.6	3.6	345	4.83
5.	C ₆ H ₅ -	C ₆ H ₅ CO-	6.0	2.9	2-benzoyl-4,5-dioxo-phenyl-7-(phenylamino)-pyranodioxin	41	180	C ₂₆ H ₁₇ NO ₆	71.0	3.7	3.2	71.1	3.9	3.2	351	4.29
6.	CH ₃ -	n-C ₄ H ₉ -	6.5	3.7	2-butyl-2-methyl-4,5-dioxo-7-(phenylamino)-pyranodioxin	73	161	C ₁₈ H ₁₉ NO ₅	65.8	5.6	4.2	65.6	5.7	4.2	343	4.55
7.	C ₂ H ₅ -	n-C ₃ H ₇ -	6.5	3.7	2-ethyl-4,5-dioxo-7-(phenylamino)-2-propyl-pyranodioxin	77	154	C ₁₈ H ₁₈ NO ₅	65.2	5.5	4.0	65.6	5.7	4.2	343	4.18
8.	n-C ₃ H ₇ -	n-C ₃ H ₇ -	7.0	3.7	4,5-dioxo-7-(phenylamino)-2,2-dipropyl-pyranodioxin	62	172	C ₁₉ H ₂₁ NO ₅	66.0	6.1	4.0	66.4	6.1	4.1	342	4.66

TABLE 3. FORMATION OF 2,2-DISUBSTITUTED-7-HYDROXY-4,5-DIOXO-6-PHENYL-PYRIDINO [4,3-d] [1,3] DIOXINS.

No.	7-(Phenylamino)-pyranodioxin (II)		Quantity (g)	Sodium/Phenol	7-Hydroxy-6-phenyl-pyridino [4,3-d][1,3] dioxin (III)	Yield %	m.p. C°	Molecular Formula	Analysis					
	R'	R''							Found			Required		
									C	H	N	C	H	N
1.	CH ₃ -	C ₂ H ₅ -	3.0	0.9g/30ml	20-ethyl-2-methyl-7-hydroxy-6-phenyl-pyridinodioxin	80	210	C ₁₆ H ₁₅ NO ₅	63.8	4.8	4.5	63.8	4.9	4.6
2.	CH ₃ -	C ₆ H ₅ -	3.5	0.9g/30ml	7-hydroxy-2-methyl-2,6-diphenyl-pyridinodioxin	84	124	C ₂₀ H ₁₅ NO ₅	68.8	4.1	4.4	68.7	4.9	4.0
3.	C ₂ H ₅ -	C ₆ H ₅ -	3.6	1.0g/30ml	2-ethyl-7-hydroxy-2,6-diphenyl-pyridinodioxin	42	159	C ₂₁ H ₁₇ NO ₅	69.6	4.6	3.7	69.9	4.6	3.8
4.	CH ₃ -	p-Cl-C ₆ H ₄ -	3.8	1.0g/30ml	2-(chlorophenyl)-7-hydroxy-2-methyl-6-phenyl-pyridinodioxin	57	146	C ₂₀ H ₁₄ NO ₅ Cl	62.7	3.3	3.4	62.5	3.6	3.6
5.	CH ₃ -	CH ₃ (CH ₂) ₃ -	3.5	0.9g/20ml	2-butyl-7-hydroxy-2-methyl-6-phenyl-pyridinodioxin	65	157	C ₁₈ H ₁₉ NO ₅	65.7	5.4	4.2	65.6	5.7	4.2
6.	C ₂ H ₅ -	CH ₃ (CH ₂) ₂ -	3.5	0.9g/20ml	2-ethyl-7-hydroxy-6-phenyl-2-propyl-pyridinodioxin	63	158	C ₁₈ H ₁₉ NO ₅	65.8	5.7	4.1	65.6	5.7	4.2
7.	C ₃ H ₇ -	C ₃ H ₇ -	3.5	0.9g/20ml	7-hydroxy-6-phenyl-2,2-dipropyl-pyridinodioxin	46	245	C ₁₉ H ₂₁ NO ₅	66.3	6.0	4.1	66.4	6.1	4.1

TABLE 4. UV AND IR SPECTRA OF PYRIDINO [4,3-d][1,3] DIOXIN (III)

No.	Pyridino [4,3-d][1,3]-dioxin(III)		UV Light absorption 95% MeOH		IR Absorption max cm ⁻¹ mainly for the 3-6 μ region (KBr disc)	
	R'	R''	λ _{max} (nm)	Logε	C-O(4)(cm ⁻¹)	C-O(5)(cm ⁻¹)
1.	CH ₃ -	C ₂ H ₅ -	310	3.90	1700	1650
2.	CH ₃ -	C ₆ H ₅ -	310	4.26	1700	1655
3.	C ₂ H ₅ -	C ₆ H ₅ -	312	4.21	1705	1660
4.	CH ₃	p-C ₆ H ₄ -	310	4.79	1700	1650
5.	CH ₃ -	n-C ₄ H ₉ -	310	4.06	1723	1668
6.	C ₂ H ₅ -	n-C ₃ H ₇ -	308	4.33	1702	1662
7.	n-C ₃ H ₇ -	n-C ₃ H ₇ -	310	4.29	1705	1655

nylamino)-4,5-dioxo-pyrano[4,3-d][1,3] dioxins (II) which underwent the phenoxide rearrangement to yield the corresponding 2,2-disubstituted-7-hydroxy-4,5-dioxo-6-phenylpyridino[4,3-d][1,3] dioxins (III). Thus the compound 2-ethyl-2-methyl-4,5-dioxo-7-(phenylamino) pyrano [4,3-d][1,3]dioxin (II, R'=CH₃, R''=C₂H₅) on the reaction with sodium phenoxide in phenol gave an enolic product C₁₆H₁₅NO₅ (III, R'=CH₃, R''=C₂H₅) m.p. 210°C, which dissolved in aq. NaHCO₃ and was isomeric with the starting material. Other 7-(phenylamino)-pyranodioxins (II) gave similar isomeric products (III).

These pyridinodioxins (III) absorbed characteristically in the UV region 308-312nm (Table 4). The IR absorption peak due to the ester carbonyl group at position 4 appeared at 1700-1725 cm⁻¹ which is apparently unbonded, while the peak due to carbonyl at position 5 appeared at 1650-1670 cm⁻¹. This is in concordance with the study of IR spectra of similar compounds carried out by Saleh and Hammad [13,14]. Treatment of the product (III, R'=CH₃, R''=C₄H₉) with diazomethane in chloroform having λ_{max} 308nm gave a negative FeCl₃ test.

The OH group of the pyridinodioxin (III, R'=C₂H₅, R''=C₃H₇) reacts with POCl₃ to yield the corresponding chloro derivative (IV) C₁₂H₁₃O₅Cl, m.p. 160°C having UV absorption at λ_{max} 324nm which shows a shift of 12nm towards the visible region as compared with the parent compound.

Bromination of the compound (III, R'=C₂H₅, R''=C₆H₅) with bromine in chloroform yielded the bromo derivative (V) C₂₁H₁₆NO₅Br, m.p. 124°C showing a UV absorption λ_{max} at 312nm.

These 7-hydroxy pyridinodioxins also formed addition products with morpholine. For instance, the product (III,

R'=C₂H₅, R''=C₃H₇) when reacted with morpholine gave an adduct with molecular formula C₂₂H₂₈N₂O₆, m.p. 193°C, λ_{max} 311nm which was soluble in water and which could be converted to the parent compound by the acidification of its aqueous solution.

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