## **COMPARATIVE STUDIES OF PMR SPECTRA OF SOME DIHYDROPYRIDINES. PART III**

A. KAMAL, TAHIRA BEGUM, M. AFROZE KHAN and ASAF A. QURESHI

PCSIR Laboratories, Karachi 39

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Comparative PMR studies of the 2,6-dimethyl-3,5-diethoxycarbonyl-4-(phenyl)-1,4-dihydropyridine (I), 2,6-dimethyl-3,5-diethoxycarbonyl-4-(2'-chlorophenyl) - 1,4 - dihydropyridine (II), 2,6-dimethyl-3,5-diethxy-carbonyl-4-(3'-hydroxyphenyl) - 1, 4-dihydropyridine (III), 2,6-dimethyl-3,5-diethoxycarbonyl-4-(4'-nitrophenyl) - 1,4-dihydropyridine (IV), 2,6-dimethyl-3, 5-diethoxy-carbonyl-4-(3',4'-dimethoxyphenyl)-1,4-dihydropyridine(V), 2,6-dimethyl-3, 5-diethoxycarbonyl-4-(a'-naphthyl) - 1,4-dihydropyridine (VI) are described.

In our previous communications we have described the PMR spectral studies of some isocoumarins<sup>I</sup>, dihydropyridines and trithians.<sup>2</sup> In continuation of these studies we are describing PMR studies of some more dihydropyridines available to us.

The PMR spectrum of 2-6,dimethyl-3, 5diethoxycarbonyl - 4 - phenyl - 1,4-dihydropyridine (I) showed a triplet and a quartet centred at  $\tau 8.85$  (6H) and  $\tau 5.6$  (4H) due to an A<sub>2</sub>B<sub>3</sub> system of the ethyl groups of the two ester groups, substituted at positions 3 and 5 ( $J_{A_2A_3} = 7$  Hz). The two *C*-methyl groups substituted at position 2 and 6 appeared as a sharp singlet at  $\tau$  7.6 (6H). Methine proton gave a single peak at  $\tau 4.7$  (1H). The five phenyl protons being in identical environment gave a sharp peak centred at  $\tau 2.1$  (5H). Finally, the NH proton appeared at  $\tau 0.2$  (1H).

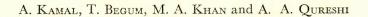
The PMR spectrum of 2,6-dimethyl-3,5diethoxycarbonyl-4-(2'-chlorophenyl),1,4-dihydropyridine (II) showed a triplet and a quartet centred at  $\tau 8.75$  (6H) and  $\tau 5.7$  (4H) due to A<sub>2</sub>B<sub>3</sub> system of the ethyls of the two ester groups, substituted at 3 and 5 positions ( $J_{A_2B_3} = 7$  Hz). The two *C*-methyl groups present at 2 and 6 positions appeared as a singlet at  $\tau$  7.6 (6H) due to identical environment. The methine proton at position 4 appeared as a singlet at  $\tau$  6.2 (1H). The four benzenoid protons appeared at  $\tau$  1.9 (4H). Finally, the NH proton appeared at  $\tau 0.5$ (1H).

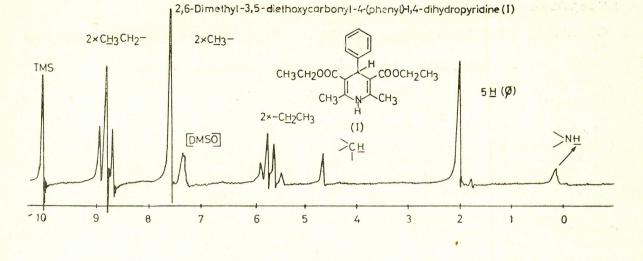
The PMR spectrum of 2, 6-dimethyl - 3, 5diethoxycarbonyl-4 - (3' - hydroxyphenyl) - 1, 4dihydropyridine (III) showed a triplet and a quartet centred at  $\tau$  8.7 (6H) and  $\tau$  5.7 (4H) respectively due to A<sub>2</sub>B<sub>3</sub> system of the ethyl of two ester groups substituted at 3 and 5 positions ( $J_{A_2B_3} = 7$  Hz). The two *C*-methyl groups present at 2 and 6 positions as a singlet at  $\tau$  7.6 (6H) due to identical environment. The methine proton at position 4 appeared as a singlet at  $\tau$  4.75 (1H). The four benzenoid protons appeared as a singlet at  $\tau$  3.0(4H). The OH proton appeared at  $\tau$  0.8 (1H). Finally, the NH proton appeared at  $\tau$  0.5 (1H).

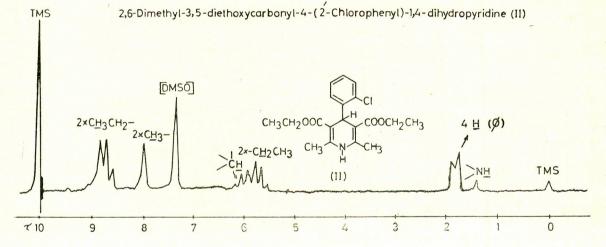
The PMR spectrum of 2,6-dimethyl-3,5diethoxycarbonyl-4- (4'-nitrophenyl)-1,4-dihydropyridine (IV) showed a triplet and a quartet centred at  $\tau$  8.75 (6H) and  $\tau$  5.8 (4H) respectively due to A<sub>2</sub>B<sub>3</sub> system of the two ethyls of the two ester groups, substituted at positions 3 and 5. ( $J_{A_2B_3} = 7$  Hz). The two C-methyl groups substituted at position 2 appeared as a sharp singlet at  $\tau$  7.55 (6H) due to identical environment. The methine proton at position 4 appeared as a singlet at  $\tau$  5.6 (1H). The four benzenoid protons appeared as two doublets centred at  $\tau$  2.12 (2H; *bb'*) and  $\tau$  1.38 (2H; *aa'*) showing *ortho* coupling (7 Hz). The *aa'* protons appeared at lower  $\tau$ value due to the neighbouring NO<sub>2</sub> group. The proton of NH appeared at  $\tau$  0.5 (1H).

The PMR spectrum of 2,6-dimethyl -3,5diethoxycarbonyl-4-(3', 4'-dimethoxyphenyl) -1,4dihydropyridine (V) showed a triplet and a quartet centred at  $\tau$  8.75 (6H) and  $\tau$  5.7 (4H) due to A<sub>2</sub>B<sub>3</sub> system of the ethyls of two ester groups, substituted at the 3 and 5 positions ( $J_{A_2B_3} = 7Hz$ ). The two *C*-methyl groups present at 2 and 6 positions appeared as a singlet at  $\tau$  7.6 (6H) being in identical environment. The two —OCH<sub>3</sub> groups being in identical environment appeared as a sharp singlet at  $\tau$  6.05 (6H). The methine proton at position 4 appeared as a singlet at  $\tau$  4.82. The three benzenoid protons appeared as a doublet centred at  $\tau$  2.67 (3H) showing *meta* coupling (J=3 Hz). Finally, the NH proton appeared at  $\tau$  0.5. (1H).

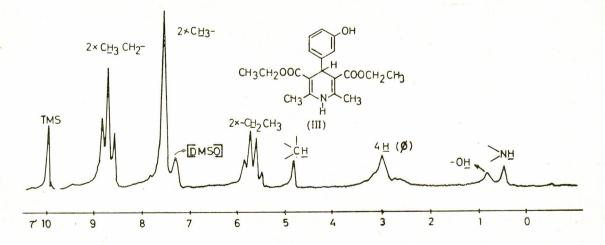
The PMR spectrum of 2,6-dimethyl-3,5diethoxycarbonyl-4-( $\alpha$ -naphthyl)-1,4-dihydropyridine (VI) showed a triplet and a quartet centred at  $\tau$  9.1 (6H) and  $\tau$  5.7 (4H) due to A<sub>2</sub>B<sub>3</sub> system of the ethyls of the two ester groups, substituted at positions 3 and 5 ( $J_{A_2B_3} = 7$  Hz). The two C-methyls substituted at positions 2 and 6 appeared as a sharp singlet at  $\tau$  7.52 (6H).





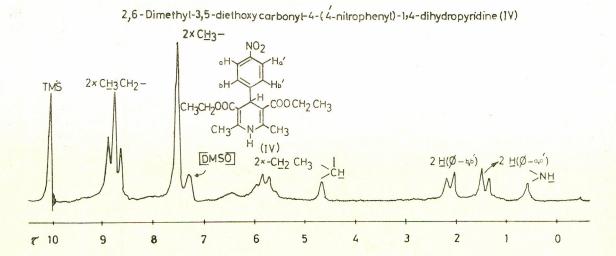


2,6-Dimethyl-3,5-diethoxycarbonyl-4-(3-hydroxyphenyl) -1,4-dihydropyridine (III)

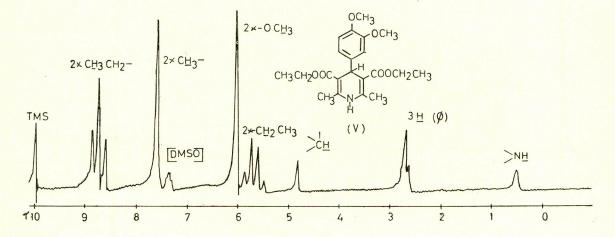


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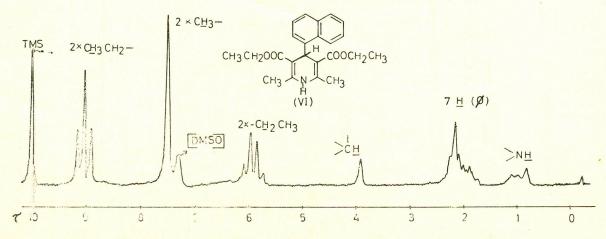
PMR Spectra of Dihydropyridines. Part III



2,6-Dimethyl-3,5-diethoxycarbonyl-4-(3,4-dimethoxyphenyl)-1,4-dihydropyridine (V)



2,6-Dimethyl-3,5-diethoxy carbonyl-4- (-- naphthyl)-1,4-dihydropyridine (VI)



Methine proton gave a sharp peak at  $\pm 3.9$  (1H), this lower value is due to the presence of the neighbouring naphthyl group. The seven naphthyl protons appeared between  $\pm 1.7 \sim \pm 2.4$  (7H). The proton of the >NH group gave a singlet at  $\pm 0.8$ (1H), thus accounting for all the protons in the molecule.

## Experimental

All the compounds were run on DP-60 Varian Associates NMR machine in DMSO, using TMS as an internal standard. All the compounds were run in DMSO (d<sub>6</sub>) due to their low solubility in other solvents. The concentration of the each compound was about 11-15%.

## References

- A. Kamal, Nilofer Kazi, Tahira Begum, M. Afroze Khan and Asaf A. Qureshi, Pakistan J. Sci. Ind. Res., 14, 1 (1971).
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  A. Kamal, Rafia Akhtar, Tahira Begum, Asaf A. Qureshi and M. Afroze Khan, Pakistan J. Sci. Ind. Res., **14**, 6 (1971).